Rajiv Gandhi University of Health Sciences, Karnataka

I Year Pharma-D Examination - Mar 2013

Time: Three Hours Max. Marks: 70 Marks

PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. CODE: 2854

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. a) Give an account on Markonikov's and Anti Markonikov's additions to alkenes, giving examples.
 - b) Explain the mechanism and orientation involved in the hydration of alkenes
- 2. Give the method of preparation and uses of the following
 - a) Benzyl benzoate b) Lactic acid c) Saccharin sodium d) Methyl salicylate e) Dimercaprol
- 3. a) Discuss the resonance and orbital descriptions of Allyl cation.
 - b) Explain hyperconjugation and its importance in the study of stability of carbocations and free radicals.

SHORT ESSAYS (Answer any Six)

 $6 \times 5 = 30 \text{ Marks}$

- 4. Discuss the mechanism of Friedel-Craft's alkylations in benzene. What are their limitations?
- 5. Write a note on the acidity of carboxylic acids.
- 6. Discuss the conditions that favour unimolecular substitution over bimolecular substitution in Alkyl halides. https://www.rguhsonline.com
- 7. Write a note on Heat of hydrogenation and compare the stability of alkenes.
- 8. Explain the concept of aromaticefy and Huckel's rule with examples.
- 9. Define and classify isomerism with examples.
- 10. Write a note on Elimination Vs Substitution.
- 11. Explain the orientation and rearrangements involved in E_1 reactions.

SHORT ANSWERS $10 \times 2 = 20 \text{ Marks}$

- Draw the structural formula for the following:a)trans-1,2-dichloroethene b) 1-Penten-4-yne
- 13. Explain the term "Polarity of bonds".
- 14. What is crossed Aldol condensation? Give the equation.
- 15. How will you assay dimercaprol?
- 16. Draw the structural formula for the following:
 - a) 2-Methyl-2butene b) 2,5-Dimethyl-2-hexene.
- 17. Arrange the free radicals in their order of stability.
 - a) Vinyl, CH₂=CH b) Allyl, CH₂=CH-CH₂ c) Benzyl, C₆H₆-CH₂
- 18. Give the Lowry-Bronsted and Lewis theory of acid and base.
- 19. Give the structure and uses of
 - a) Chlorbutol b) Tartaric acid
- 20. Classify amines with examples and structures.
- 21. Distinguish between intra and intermolecular hydrogen bonding. Give examples.
